



Synthesis of the Perhydroindole Nucleus by a Pummerer/Mannich Induced Cyclization Cascade

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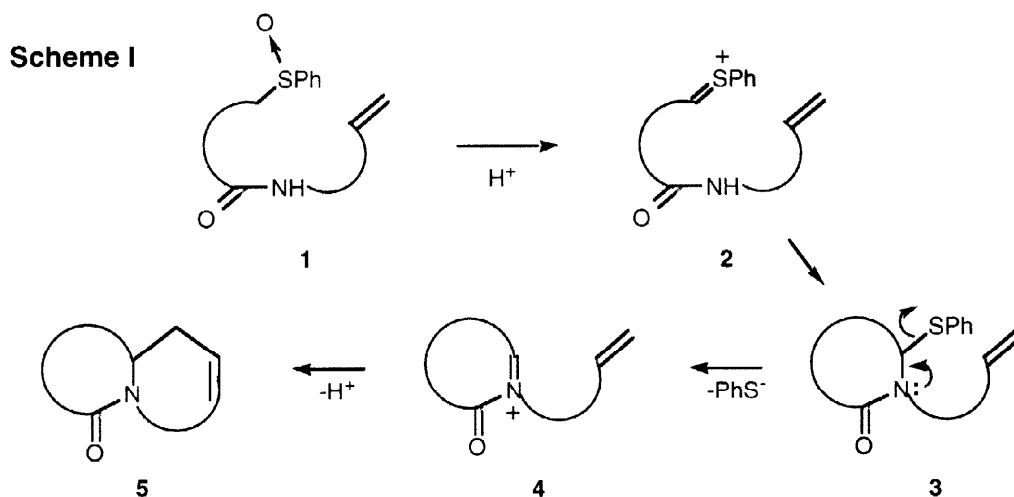
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Abstract: The silicon-induced Pummerer reaction of several amido sulfoxides possessing tethered π -bonds proceeds by way of a thionium/*N*-acyliminium ion cascade to provide various azabicyclic ring systems. This cascade sequence was applied toward the synthesis of a member of the protoberberine alkaloid family. © 1998 Elsevier Science Ltd. All rights reserved.

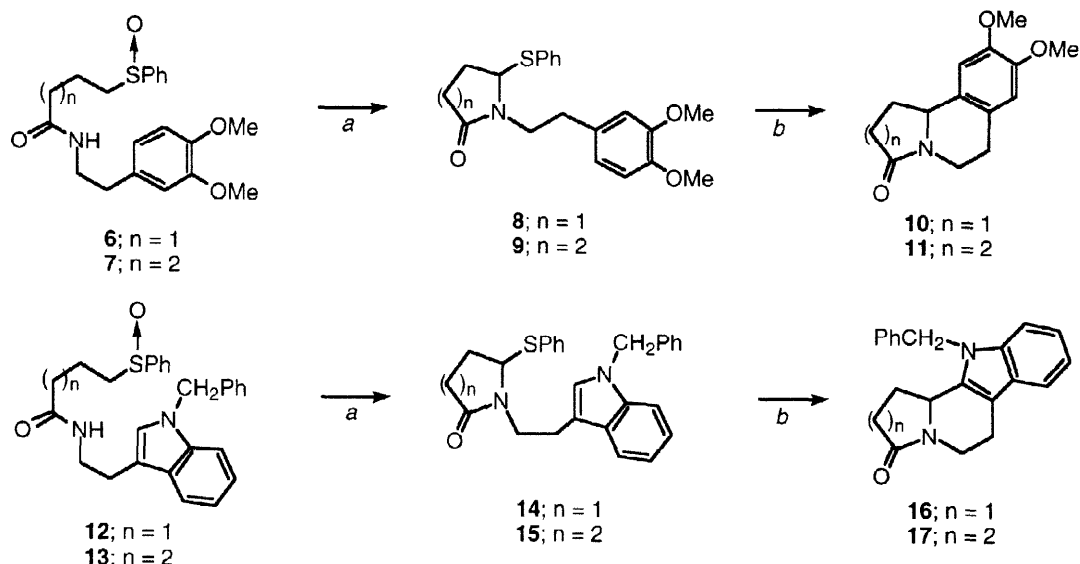
The indolizidine and quinolizidine alkaloids are natural products which have been isolated from plants, fungi, and animal sources.¹ They have aroused considerable interest because of the potent and useful biological activity of certain of its members,^{2,3} especially the polyhydroxy derivatives which function as inhibitors of several glycosidases.⁴ The common structural feature of these compounds is a six-membered nitrogen heterocycle incorporated into a bicyclic ring. The development of new general methods for the synthesis of these heterocycles remains an area of active investigation.^{5,6} An exceptionally viable strategy that has been utilized for the preparation of many five and six-membered nitrogen heterocycles involves the addition of nucleophiles to *N*-acyliminium ions.⁷ In particular, the intramolecular reaction of cyclic *N*-acyliminium ions has been successfully employed in the preparation of various azabicyclic ring systems found in natural products.⁸

Our interest in indolizidine/quinolizidine alkaloid synthesis was prompted by the desire to explore the *thionium/N*-acyliminium ion cascade as a key strategy for the assembly of these ring systems. The approach we had in mind was based on our previous success using a tandem Pummerer/Mannich cyclization sequence for synthesis of the erythrinane alkaloid skeleton.⁹ We envisioned that thionium ion



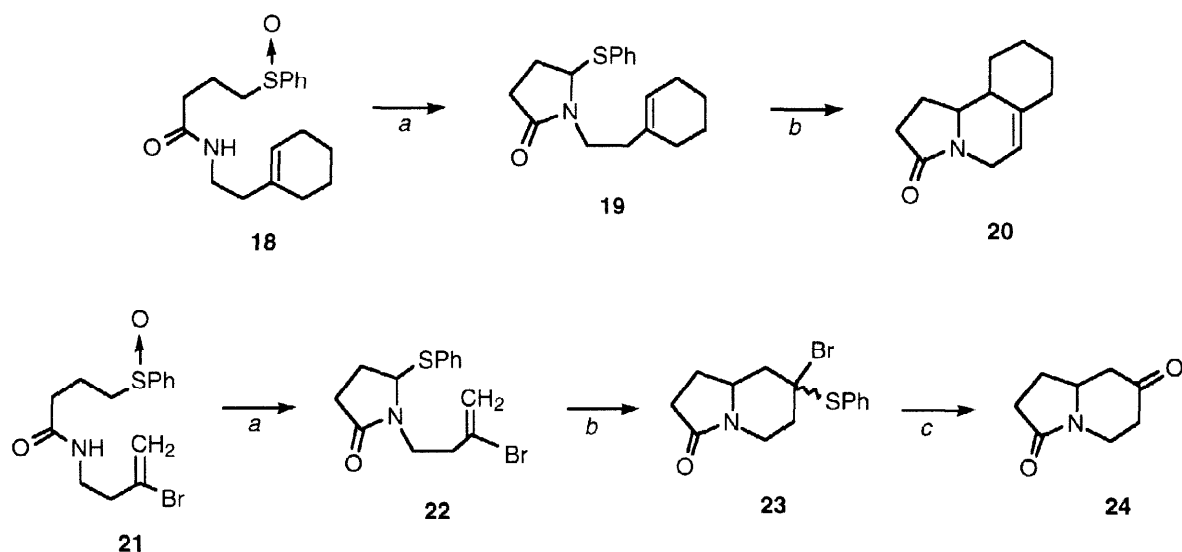
2, derived from a Pummerer reaction of sulfoxide 1, would readily react with the neighboring amido nitrogen atom to provide the 2-thiophenyl substituted lactam 3.¹⁰ Subsequent elimination of the thiophenyl group should ultimately lead to the azabicyclic lactam 5 via cyclization of a transient *N*-acyliminium ion (*i.e.*, 4). The present communication documents the results of our studies in this area.

Amido sulfoxides **6** and **7** were easily prepared by addition of thiophenol to the appropriate alkenoic acid π -bond, and this was followed by reaction of the *in situ* generated acyl chloride with 3,4-dimethoxyphenethylamine. The silicon-induced Pummerer reaction of these amido sulfoxides was carried out using 1-(dimethyl-*tert*-butylsiloxy)-1-methoxyethylene in dry acetonitrile in the presence of a catalytic amount of ZnI_2 as described by Kita¹¹ and led to the very clean formation (>90%) of 2-thio substituted lactams **8** and **9**. Iminium ion-aromatic π -cyclization was readily accomplished by treatment of **8** or **9** with 1.2 equiv of $\text{BF}_3 \cdot 2\text{AcOH}$ in CH_2Cl_2 at 25 °C to provide bicyclic lactams **10** or **11** in 98% and 79% yield, respectively. A related set of reactions occurred using the indolyl substituted amido sulfoxides **12** and **13** which afforded indoles **16** and **17** in excellent yield from the initially formed Pummerer products **14** and **15**.



Reagents: (a) TBDMSOC(OMe)=CH_2 , ZnI_2 ; (b) $\text{BF}_3 \cdot 2\text{AcOH}$

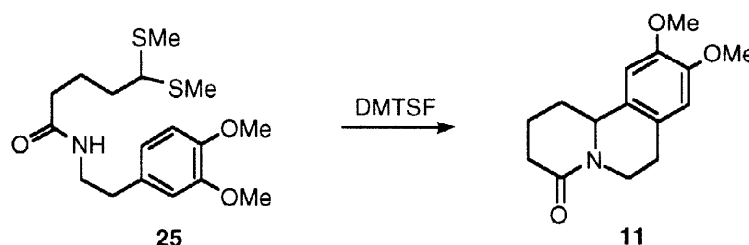
Since the previous examples involve aromatic π -bond cyclization, we decided to study several systems which possess a simple olefinic tether. We found that treatment of the cyclohexenyl substituted amidosulfoxide **18** with the *t*-butyl O-silylated ketene acetal caused an intramolecular Pummerer-type



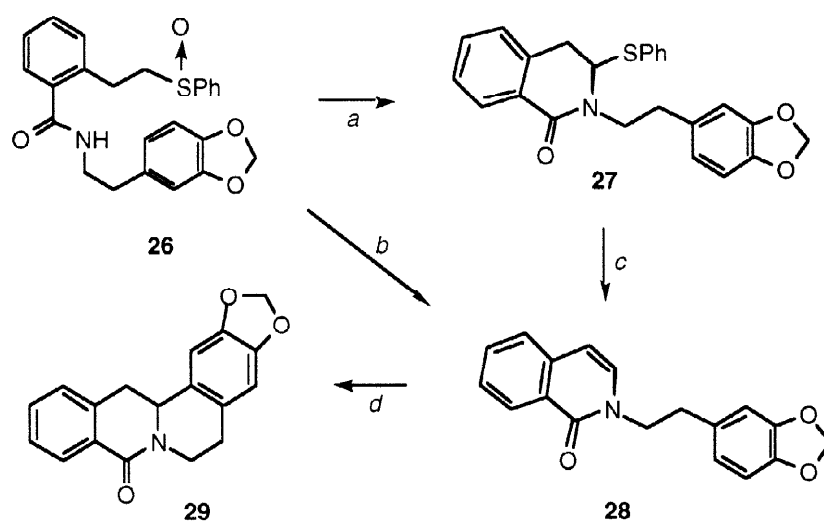
Reagents: (a) TBDMSOC(OMe)=CH_2 , ZnI_2 ; (b) $\text{BF}_3 \cdot 2\text{AcOH}$; (c) Hg(OAc)_2 , HCO_2H

reaction to give α -thiolactam **19** which was subsequently converted to **20** upon treatment with $\text{BF}_3 \cdot 2\text{AcOH}$ in 50% overall yield. Extension of the two-step sequence to the bromoalkenyl substituted amide **21** was investigated next. When **21** was subjected to the typical Kita Pummerer conditions,¹¹ the desired 2-thiolactam **22** was isolated in 80% yield. Further reaction of **22** with $\text{BF}_3 \cdot 2\text{AcOH}$ furnished the novel bromo-thiophenyl substituted indolizidone **23** which was hydrolyzed to ketolactam **24** in good yield.

Initial attempts to induce a "one-pot" tandem cascade of the starting amidosulfoxides failed to produce the cyclized product. However, the desired domino cascade sequence could be induced by using a dithioacetal substituted amide as the thionium ion precursor. Thus, treatment of amide **25** with dimethyl-(methylthio)sulfonium fluoroborate (DMTSF), as described by Trost,¹² initiated a one-pot cascade sequence to deliver hexahydroisoquinolinone **11** in near quantitative yield.



Our interest in establishing amidosulfoxides as useful building blocks for heterocyclic synthesis prompted us to use the Pummerer methodology for the preparation of a member of the protoberberine alkaloid family.¹³ The protoberberines are a large class of natural products typically characterized by a tetracyclic ring skeleton with an isoquinoline core. Considerable efforts have been expended in the study of these molecules for both their synthetic and biological significance.¹³ Most of the synthetic approaches are generally plagued by the non-availability of starting materials, multi-step procedures, and moderate to poor yields.¹⁴ A short synthesis of the berberine derivative **29** was carried out as depicted below. Subjection of amidosulfoxide **26** to $\text{TMSOTf}/\text{NEt}_3$ as the Pummerer initiator afforded 2-thiophenyl lactam **27** in 64% yield. In contrast, when the Kita silicon conditions¹¹ were used to trigger the Pummerer reaction, only enamide **28** (85% yield) was obtained. The reaction of **27** with Lewis acids such as ZnI_2 did result in the formation of **28**. When **28** was exposed to acidic conditions, it was transformed into **29** in 55% yield (unoptimized).¹⁵



Reagents: (a) $\text{TMSOTf}/\text{NEt}_3$; (b) TBDMSOC(OMe)=CH_2 , ZnI_2 ; (c) ZnI_2 ; (d) H^+

In conclusion, this study has demonstrated that the *thionium/N-acyliminium ion cyclization sequence* of amido sulfoxides represents a highly efficient method for the synthesis of azabicyclic ring systems. The further utilization of this cyclization cascade for the stereocontrolled synthesis of perhydroindole alkaloids is under current investigation.

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References and Notes

- Howard, A. S.; Michael, J. P. In *The Alkaloids*; Brossi, A., Ed.; Academic Press: New York, 1986; Vol. 28, p 183. Michael, J. P. *Nat. Prod. Rep.* **1994**, *11*, 17.
- Fodor, G. B.; Colasanti, B. In *Alkaloids: Chemical and Biological Perspectives*; Pelletier, S. W., Ed.; John Wiley and Sons: New York, 1985; Vol. 3, p 1. Strunzand, G. M.; Findlay, J. A. In *The Alkaloids*; Brossi, A., Ed.; Academic Press: New York, 1986; Vol. 26, p 89. Numata, A.; Ibuka, T. In *The Alkaloids*; Brossi, A., Ed.; Academic Press: New York, 1987; Vol. 31, p 193.
- Daly, J. W.; Garraffo, H. M.; Spande, T. F. In *The Alkaloids*; Cordell, G., Ed.; Academic Press: San Diego, 1993; Vol. 43, p 185. Elbein, A. D.; Molyneux, R. J. In *Alkaloids: Chemical and Biological Perspectives*; Pelletier, S. W., Ed.; Springer Verlag: New York, 1986; Vol. 4, p 1.
- Wong, C. H.; Halcomb, R.; Ichibaka, Y.; Kajimoto, T. *Angew. Chem., Int. Ed. Engl.* **1995**, *34*, 521. Sinnott, M. L. *Chem. Rev.* **1990**, *90*, 1171.
- Michael, J. P. *Nat. Prod. Rep.* **1995**, *12*, 535; *ibid* **1993**, *10*, 639.
- Angle, S. R.; Breitenbucher, J. G. In *Studies in Natural Products Chemistry; Stereoselective Synthesis*, Atta-ur-Rahman, Ed.; Elsevier: New York, 1995; Vol. 16, Part J, p 453. Comins, D. L.; Zhang, Y. M. *J. Am. Chem. Soc.* **1996**, *118*, 12248. Munchhof, M. J.; Meyers, A. I. *J. Am. Chem. Soc.* **1995**, *117*, 5399. Sisti, N. J.; Zeller, E.; Grierson, D. S.; Fowler, F. W. *J. Org. Chem.* **1997**, *62*, 2093. Pearson, W. H.; Hembre, E. J. *J. Org. Chem.* **1996**, *61*, 5537.
- Hiemstra, H.; Speckamp, W. N. In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I., Eds.; Pergamon: Oxford, 1991; Vol. 2, pp 1047-1082. Hart, D. J. In *Alkaloids: Chemical and Biological Perspectives*; Pelletier, S. W., Ed.; Wiley: New York, 1988; Vol. 6, p 227.
- Flann, C.; Overman, L. E. *J. Am. Chem. Soc.* **1987**, *109*, 6115. Heitz, M. P.; Overman, L. E. *J. Org. Chem.* **1989**, *54*, 2591. Gramain, J. C.; Remuson, R. *Tetrahedron Lett.* **1985**, *26*, 4083.
- Padwa, A.; Hennig, R.; Kappe, C. O.; Reger, T. S. *J. Org. Chem.* **1998**, *63*, 1144.
- For a review describing the intramolecular addition of heteroatoms to Pummerer intermediates, see: Padwa, A.; Gunn, D. E., Jr.; Osterhout, M. H. *Synthesis* **1997**, 1353.
- Kita, Y.; Yasuda, H.; Tamura, O.; Itoh, F.; Tamura, Y. *Tetrahedron Lett.* **1984**, *25*, 4681. Kita, Y.; Tamura, O.; Miki, T.; Tamura, Y. *Tetrahedron Lett.* **1987**, *28*, 6479.
- Trost, B. M.; Murayama, E. *J. Am. Chem. Soc.* **1981**, *103*, 6529.
- Bkakuni, D. S.; Jain, S. "Protoberberine Alkaloids" In *The Alkaloids: Chemistry and Pharmacology*; Brossi, A., Ed.; Academic Press: Orlando, 1985; Vol. 26, p 229.
- Matulenko, M. A.; Meyers, A. I. *J. Org. Chem.* **1996**, *61*, 573.
- All new compounds in this study were fully characterized (IR, NMR, elemental analysis and/or HRMS).